# Synthesis, characterization and study of methyl 3-(2-oxo-2H-1,4-benzoxazin-3-yl) propanoate as new corrosion inhibitor for carbon steel in $1 \mathrm{M} \mathrm{H}_{2} \mathrm{SO}_{4}$ solution 

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#### Abstract

A new benzoxazin derivative, namely methyl 3-(2-oxo-2H-1,4-ben-zoxazin-3-yl) propanoate (1,4-MBXP), was synthesized under mild conditions from 2-aminophenols and dimethyl-2-oxoglutarate. The prepared compound was identified by FT-IR, ${ }^{1} \mathrm{H}$ NMR, ${ }^{13} \mathrm{C}$ NMR spectroscopies, elementary analysis and MS. Its inhibitive action against the corrosion of carbon steel in $1 \mathrm{M} \mathrm{H}_{2} \mathrm{SO}_{4}$ solution was investigated by weight-loss and hydrogen evolution measurements. $1,4-\mathrm{MBXP}$ is a good corrosion inhibitor and its inhibition efficiency increases with the increase of concentration to attain $75.08 \%$ at 180 ppm . The temperature effect on the corrosion behaviour of carbon steel in $1 \mathrm{M} \mathrm{H}_{2} \mathrm{SO}_{4}$ with and without the inhibitor at 180 ppm was studied in the temperature range from 298 to 338 K . The synergistic action caused by iodide ions on the corrosion inhibition of carbon steel in $1 \mathrm{M} \mathrm{H}_{2} \mathrm{SO}_{4}$ by 1,4-MBXP at 180 ppm was studied using weight methods at 298 K . The inhibition efficiency synergistically increased on addition of potassium iodide.


Keywords Corrosion inhibition • 1,4-benzoxazin • Carbon steel • Weight-loss • Adsorption • Sulfuric acid

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